

Ford: Chs 1-3









### Ch. 1

- Atoms mostly empty space (e.g. propeller)
- Atoms very big compared to subatomic world
- Existence of atoms still in doubt in early 1900s
- Some particles: electron, proton, photon
- Standard Model: quarks, Higgs, leptons, force carriers
- · Classical determinism vs quantum probabilities

# Ch. 2—Small/Fast

- Scientific Notation (of necessity)
- Range of units (eV, angstroms, etc)
- Physical Constants (h, hbar, c)
- What does it mean to be quantized?
- Charge:
  - Fundamental (property & force)
  - We're not sure why it's quantized
  - Holds electrons in atom
  - Balances strong force (gluons) inside nucleus
- Spin
  - A measure of angular momentum (different than orbital)
  - Fundamental property of particles (half-integer, and integer)
  - quantized





Structure of re-8 molecular nanomagnets and SQUID nighfrequency EPR spectra of Fe-8 showing resonant absorption corresponding to transitions between the quantized energy levels.

#### From:

http://www.boulder.nist.gov/div818/81803/2005/MagneticThinFilms/index.html







# Ch. 3—The Leptons Antiparticles/antimatter—particles identical to their partners, aveant the approximate charge (same)

- their partners except the opposite charge (same mass and spin)
- Electron (1897 JJ Thompson)
  - First fundamental particle discovered
  - Basis of all electronics (but for how much longer?)

• Leptons

Neutrinos

- 3 flavors
- Only weakly interacting—no charge
- Hard to detect
- Very little masses



## Muon

- 200x more massive than electron
- Otherwise very similar to electron
- Decays

### Tau

- Very massive (more than proton)
- Same charge as muon and electron